

Name:

Date:

Pd:

# Metric and Chemistry Conversions Practice

## Chemistry Conversion Factors:

1 mole =  $6.022 \times 10^{23}$  atoms/molecules

1 mole = 22.4 L of gas



Common Prefixes Used with SI Units			
Prefix	Symbol	Conversion Factor to Base Unit	Order of Magnitude
Giga-	G	1,000,000,000 base = 1 Giga	$10^9$
Mega-	M	1,000,000 base = 1 Mega	$10^6$
kilo-	k	1,000 base = 1 kilo	$10^3$
hecto	h	100 base = 1 hecto	$10^2$
deka-	da	10 base = 1 deka	$10^1$
	Base Unit	1 base	$10^0$
deci-	d	1 base = 10 deci	$10^{-1}$
centi-	c	1 base = 100 centi	$10^{-2}$
milli-	m	1 base = 1,000 milli	$10^{-3}$
micro-	$\mu$	1 base = 1,000,000 micro	$10^{-6}$
nano-	n	1 base = 1,000,000,000 nano	$10^{-9}$
pico-	p	1 base = 1,000,000,000,000 pico	$10^{-12}$

Complete the conversions below with a partner.  
Show all work and round to correct sig figs  
w/units:

$1.9 \times 10^{-3}$ Hz = _____ GHz	1.25 mol of O <sub>2</sub> = _____ L of O <sub>2</sub>	8.43 km = _____ meters
$7.54 \times 10^{19}$ atoms Mg = _____ mol Mg	1.5 Gigabyte (GB) = _____ bytes (Bytes is a base unit)	87.54 L of Methane = _____ mol Methane
8.65 L of CO <sub>2</sub> = _____ atoms CO <sub>2</sub>	$1.23 \times 10^{25}$ microliters = _____ L	*9.54 GHz = _____ mHz (millihertz)

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**Team Quiz:**

- *Show all work and round your final answer to two decimal places with units:*

1.	2.	3.
4.	5.	6.

**Reflect:**

*Write a short paragraph below describing how to do ANY dimensional analysis conversion:*