

Name:

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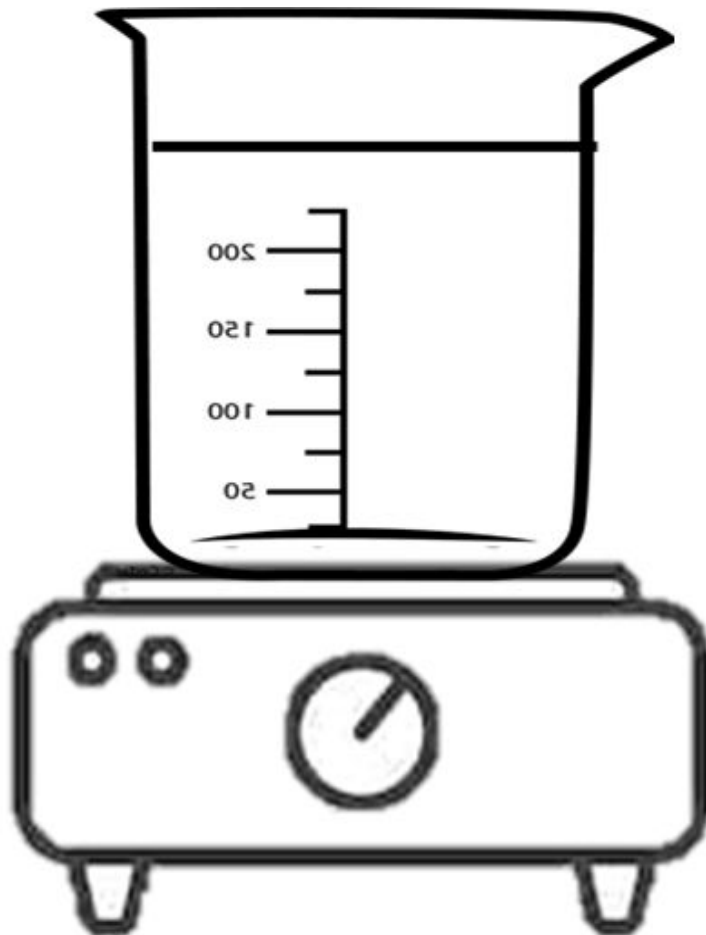
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Unit 3, Section 1 - Energy Forms and Electromagnetic Radiation - HONORS

Energy Transfer:

_____ is the process in which energy is transferred from object to another. This process can be broken down into three different methods: conduction, convection, and radiation. _____ is the first method of which energy is directly transferred from one object to another. The second method, _____, is when energy is transferred due to the presence and movement of fluids or gases. The third method, _____, is the process in which energy is transferred through waves, or more specifically electromagnetic waves. It is important to note that this method of energy transfer **does not** require the objects to be in direct contact.

Label and color the diagram below to show conduction, convection, and radiation:



Types of Energy:

| Kinetic Energy | Potential Energy |
|----------------|------------------|
| | |

Forms of Energy:

| Energy Form | Type of Energy (Kinetic or Potential) | Description |
|-------------|---------------------------------------|-------------|
| | | |
| | | |
| | | |
| | | |
| | | |
| | | |
| | | |

Practice:

Identify the type of heat transfer as conduction(CD), convection(CV), or radiation(R):

_____ The heat you feel from a fireplace

_____ Moves as a wave

_____ Transfer through solids

_____ Moves as a current

_____ A pan heating on a hot stove

_____ Sun rays on Earth

Match the form of energy with its definition.

1. _____ Heat

A. Energy of an atom being split or fused

2. _____ Nuclear

B. Energy of moving electrons (charged particles)

3. _____ Radiant

C. Energy of motion

4. _____ Mechanical

D. Light energy - electromagnetic radiation

5. _____ Chemical

E. Energy (kinetic or potential) of **moving** objects

6. _____ Electrical

F. Energy of bonds in molecules and compounds

7. _____ Potential

G. Stored energy

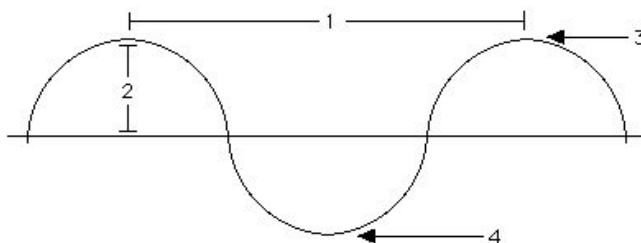
8. _____ Kinetic

H. Thermal energy - motion of molecules

Electromagnetic Radiation (Radiant Energy):

As a form of energy, light (electromagnetic radiation) travels in _____ through the environment. Below are the parts of a wave:

- 1 - _____
- 2 - _____
- 3 - _____
- 4 - _____



Energy of a wave can be identified in two easy ways: wavelength and frequency of the wave. Wavelength is a measurement of the distance from _____ to _____ on two consecutive waves and is often measured in meters or nanometers. Frequency refers to the number of waves that pass a point per _____, measured in Hertz (Hz).

Let's practice some metric conversions!

- Convert from 34 cm to meters:

| | |
|---------------------|--------------------|
| 34 centimeters (cm) | 0.01 meters (m) |
| _____ | 1 centimeters (cm) |

- Convert from 1,340 km to millimeters:

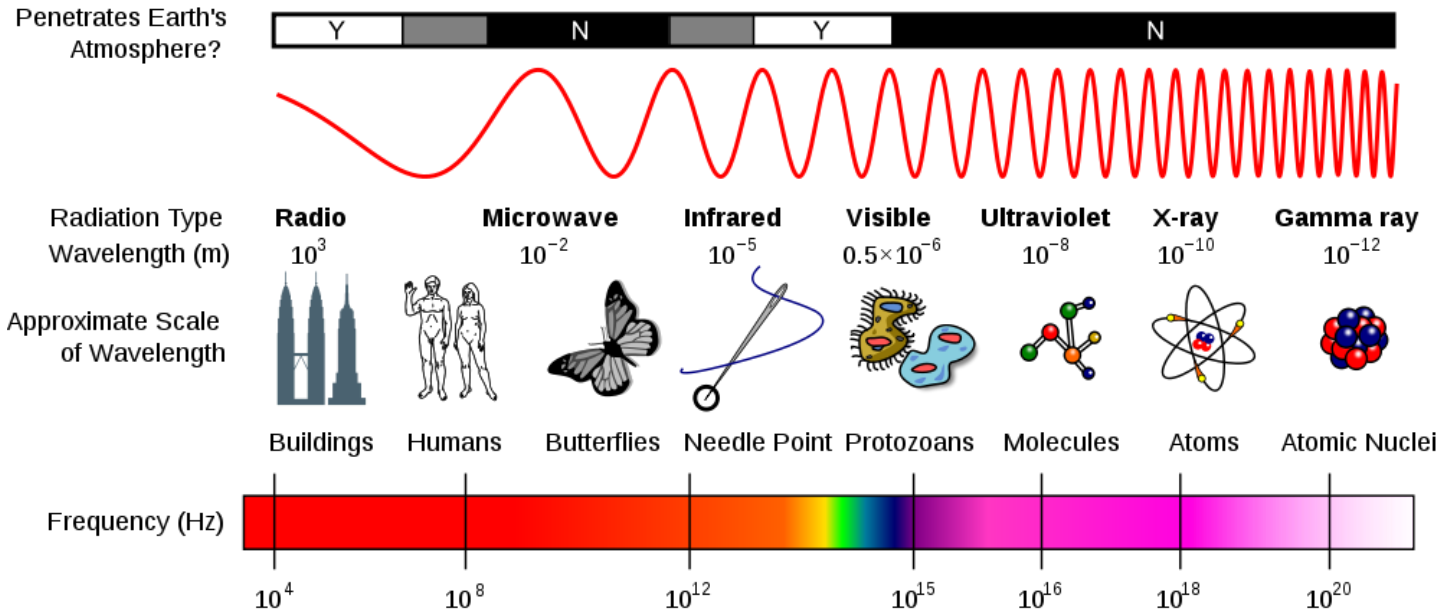
- Convert from 1.23×10^{14} nanometers (nm) to meters:

- Convert from 2.3 Megahertz (MHz) to Hertz (Hz):

| Common Prefixes Used with SI Units | | | |
|------------------------------------|-----------|---------------------------------|--------------------|
| Prefix | Symbol | Conversion Factor to Base Unit | Order of Magnitude |
| Giga- | G | 1,000,000,000 base = 1 Giga | 10^9 |
| Mega- | M | 1,000,000 base = 1 Mega | 10^6 |
| kilo- | k | 1,000 base = 1 kilo | 10^3 |
| hecto | h | 100 base = 1 hecto | 10^2 |
| deka- | da | 10 base = 1 deka | 10^1 |
| | Base Unit | 1 base | 10^0 |
| deci- | d | 1 base = 10 deci | 10^{-1} |
| centi- | c | 1 base = 100 centi | 10^{-2} |
| milli- | m | 1 base = 1,000 milli | 10^{-3} |
| micro- | μ | 1 base = 1,000,000 micro | 10^{-6} |
| nano- | n | 1 base = 1,000,000,000 nano | 10^{-9} |
| pico- | p | 1 base = 1,000,000,000,000 pico | 10^{-12} |

Electromagnetic Spectrum:

The electromagnetic spectrum is the full spectrum of all light energy. The spectrum is designed based on decreasing _____ and increasing _____. The shorter the wavelength, the _____ the energy of the wave.



Circle the correct answer for the statements/questions below:

- The waves to the RIGHT on the spectrum are at a (**higher energy** / **lower energy**) than the waves to the left.
- Which of the following energies has the LONGER wavelength? **Radio** or **Infrared**
- Which of the following energies has the SHORTER wavelength? **X-Ray** or **Microwave**

Match the following wavelengths/frequencies of light with their correct type of radiation:

- Wavelength of 1.0×10^{-5} meters (m) = _____
- Wavelength of 9.43×10^{-10} meters (m) = _____
- Frequency of 1.22×10^5 meters (m) = _____
- Frequency of 5.4×10^{15} meters (m) = _____

Now, let's put it all together. Convert the following, then identify the correct type of radiation:

- 49 nanometers (nm) = _____ meters (m) - _____
- 0.0032 nanometers (nm) = _____ meters (m) - _____